2.0.0 SIMPLE CLASSIFICATION OF SUBSTANCES ANS

For Examiners use only.

Question	Maximum Score	Candidates Score
1 - 10	57	

1.						
	(a)	(i)	100	accept answers from 98 to 100	1 (L5)	
		(ii)	to condense the water vapour	accept 'to change the gas into liquid' or 'condensation' or 'condenser' accept 'to cool the vapour into water' do not accept 'to cool the vapour or water'	1 (L5)	
		(iii)	distillation or distilling		1 (L5)	
	(b)	(i)	evaporating or boiling		1 (L6)	
			melting		1 (L6)	
			freezing		1 (L6)	
			condensing		1 (L6)	
		(ii)	A	accept 'boiling'	1 (L6)	
			D	accept 'condensing'	1 (L6)	
				accept the letters written in the correct places on the diagram		
				correct places on the diagram		[9]

2.

- (a) does not dissolve in solvent / interfere with results owtte (1) [1]
- (b) 1 and 3 (1) [1]
- (c) sample 4 (1)

two spots present (1) [2]

(d) to show position of the acids / spots (1) [1]

3.						
	(a)	lce v	varms up. Temperature rises	from -10 to 0° C. No change of state [1m]	
	(b)	Tem _l wate		. Change of state occurs. Ice changes	to liquid [1m]	
	(c) (d)		perature remains constant [1	nanges from 0°C to 100°C [1m] m]. Change of state occurs. Boils to s	team at 100ºC [Total 6m]	
4.						
т,	Ans					
(a)		liquid	precaution night overflow			
		& ignit	eor vapour might ignite WO VALID COMMENTS		2	
(b) (i		solidif exothe	yingrmie	1 1 1		
(ii) ii)	37 ℃ single		1		
(i				1	6	
				To	tal 8	
5.		40				
	(a)	(i) <i>m</i>	agnesium + oxygen ® magne	do not accept formulae	2	
		(ii) ar	ny one from	do not accept 'air' for oxygen	1	
		•	the oxygen had mass			
		· oxygen was added to the magnesium				
				accept 'magnesium has gained an element'		
		•	the magnesium has reacted	d with oxygen		
				accept 'magnesium is now part of a compound'		
	(b)	oxygen		accept 'O2'	1	
	(c)	zinc oxi	de	accept 'ZnO'	1	

-	- 1	١
1	а	П
١	u	п

	chemical change	physical change
Α	\checkmark	
В	\checkmark	
С		✓

1

all three ticks are required for the mark

[6]

6.

(a) Increase the potential energy of the molecules OR do work in separating the molecules 1 against intermolecular forces/bonds 1 (2) (b) Molecules are moving around randomly 1

spread in all directions 1 (2)

[Total 4]

1

[6]

7.

(a)	(i)	iron	do not accept '1540°C'	1
	(ii)	mercury	do not accept '-37°C'	1
(b)	solid	to a liquid	answers must be in the correct order both answers are required for the mark	1
(c) (d)	5 (i)	sodium		1 1

8.

64

(ii)

gold

 30^{E+2} wrong/ correct charge (- $\frac{1}{2}$)

9.

(a)	(i)	С		1
	(ii)	D		1
	(iii)	A and B	answers may be in either order	1
			both answers are required for the mark	
	(iv)	A and D	answers may be in either order	1
			both answers are required for the mark	
	(v)	С		1
(b)	(i)	the same	accept 'seven'	1

(ii) a random, mixed arrangement of both types of molecule should be drawn with the molecules not touching each other [7]

10.

- (a) Q (b) P or Q
- (c) T
- (d) R
- (e) Q

[Total 5m]