

## **PHYSICS FORM 2**

**TERM 1 2022** 

MARKING SCHEME

## SECTION A



 $P = \frac{M}{V} (do not award a mark for the formula)$ 

$$P = \frac{567g}{40 \text{ cm}^3} = 14.175 \text{ g/cm}^3 \text{ (correct substitution)}$$

 $P = 14.18g/cm^{3}$  (Answer must be given correct to 2d.p)

- 3. Volume of drop =  $5 \times 10^{-8} \text{ M}^3$ 
  - i. Area of circular film =  $0.1 M^2$

$$V = A x H$$



$$h=\frac{V}{A}\sqrt{}$$

Size of molecule =  $\frac{5 \times 10^{-8} m^2}{0.1 m^2}$ 

 $= 5.0 \text{ x} 10^{-7} \text{ m} \sqrt{10^{-7}}$ 

## Accept 0.0000005

Check for correct units.

Atoms are spherical  $\sqrt{}$ ii.

Mass uniformly distributes  $\sqrt{}$ 

4. Weight on Earth = 600N

Weight on Planet = 450NWeight, W = Mg $M = \frac{W}{g}$ g√

Mass of body = 
$$\frac{600N}{10N/Kg} = 60Kg$$

$$g = \frac{w}{m}$$

$$g = \frac{450N\sqrt{}}{60Kg} = \frac{7.5N}{Kg}\sqrt{}$$

Correct substitution  $\sqrt{}$ 

Correct answer with correct units  $\sqrt{}$ 



- The force of cohesion within the mercury is greater than the force of adhesion between mercury and glass √. The mercury therefore sinks down √the tube to enable mercury molecules to keep together√.
- 6. Temperature rise and impurities lower the surface tension of water  $\sqrt{}$
- 7. a)



b) The unbalanced  $\sqrt{1}$  surface tension  $\sqrt{1}$  pulls the thread tight

8. h = 760mm  $p = 1.36 \times 10^4 \text{ Kg/m}^3$  p = ? p = pgh  $p = 1.36 \times 10^4 \times 10 \times \frac{760}{1000}$ Check on the conversion  $\sqrt{}$ Correct substitution  $\sqrt{}$ 

 $P = 103,360 \text{ N/M}^2$ 

Accept P = 103,360 pa  $\sqrt{}$  check for correct units

- The external pressure (atmospheric) is lower than the internal pressure √: therefore the capillaries break √.
- The bottle with hole experiment if diagram used; check for labeling√: Procedure, observation and conclusion√.





Lowest jet has highest pressure

11. Solid – particles very close, hence low kinetic energy  $\sqrt{}$ .

Liquids – particles fairly free, moderate kinetic energy  $\sqrt{}$ 

Gases – particles very free, high kinetic energy  $\sqrt{}$ 

- 12. The metal blade conducts heat from the hand but the wood cannot  $\sqrt{}$
- 13.  $90 48 = 42^{0}\sqrt{}$ Drawing a normal

14.  $(20 \ge 0.3) + (20 \ge 0.3)\sqrt{-100}$  or  $20 \ge 0.6$ 

6 + 6 = 12 NM $\sqrt{}$ 

Check for correct units

15. Unlike poles attract while like poles repel $\sqrt{}$ 

Reject - unlike charges attract while like charges attract

Reject - unlike terms attract while like terms attract





Check for direction of field  $\sqrt{}$ 

Check for presence of the neutral zone  $\sqrt{}$ 

- 16. This is due to the influence of the Earth's magnetic field  $\sqrt{}$
- 17. Repulsion only occurs between 2 like poles√ but attraction may occur between 2 unlike poles or between a magnet and a magnetic materials√

## **SECTION B**

 Iron is a soft magnetic material it can easily acquire magnetism and can easily lose magnetism.



- iii. A North pole $\sqrt{}$ 
  - B South pole  $\sqrt{}$
- iv. Right hard grip rule $\sqrt{}$

It states that if a coil carrying current is grasped in the right hand such that the fingers point in the direction of current then the thumb points in the direction of North Pole $\sqrt{}$ .



- ii. It would cause overheating on the electromagnet  $\sqrt{}$ . This adversely affects the magnetism of the electromagnet  $\sqrt{}$ .
- 19. i. Smoke particles smoke particles are larger than air molecules and light enough to move when bombarded by air molecules  $\sqrt{}$

Lens – focuses the light from the lamp on the smoke particles, causing them to be observable

Microscope – enlarges/magnifies the smoke particles so that they are visible  $\sqrt{}$ 

ii. Smoke particles more randomly/zigzag  $\sqrt{}$ 

Air molecules bombard the smoke particles

Air molecules are in random motion

- iii. The speed of motion of smoke particles will be observed to be lighter/faster/speed increases√.
- 20. a) Principle of moments states that for a system in equilibrium, the sum of clockwise

moments must be equal to the sum of the anticlockwise moments.



30Kg

Clockwise moments = Anticlockwise moments

 $300 \ge 1.5 = X \ge 650 \sqrt{(correct substitution 1mk)}$  $\frac{450}{650} = \frac{650 \ge}{650}$ 

**6** | P a g e







Clockwise moments = Anticlockwise moments

$$F1 + F2 = 100N\sqrt{}$$
  
 $F1 + 60N = 100N$   
 $F1 = 100N - 60N$   
 $F1 = 40N\sqrt{}$ 

21. a) Mass of water =  $66.1 - 42.9\sqrt{}$ 

= 23.2g√

b) Volume =  $\frac{Mass}{Density} = \frac{23.2g}{1g/cm^3}$ 

 $= 23.2 \text{cm}^3 \sqrt{}$ 

Working must be shown

c) Volume of density bottle = volume of water Volume of bottle =  $23.2 \text{ cm}^3 \sqrt{}$ 

d) Mass of soil = 67.2 - 42.9

e) Mass of water that filled the space above the soil

$$= 82.0 - 67.2$$
  
= 14.8g  $\sqrt{}$ 

f) Volume of soil

Volume of water  $= \frac{Mass}{Density} \sqrt{}$ 

$$=\frac{14.8g}{1g/cm^3}$$

8 | P a g e



$$= 14.8 \text{cm}^3 \sqrt{}$$

Volume of soil = 23.2 - 14.8

 $= 8.4 \text{cm}^3 \sqrt{}$ 

g) The density of the soil  $= \frac{Mass}{Volume}$ 

$$=\frac{24.3}{8.4}\sqrt{}$$
$$=2.893 \text{g/cm}^3\sqrt{}$$

22. a) A – Seal and insulator  $\sqrt{}$ 

B – Zinc case  $\sqrt{}$ 

- D Carrbon rod  $\sqrt{}$
- b) Zinc case acts as a negative electrode  $\sqrt{}$
- c) i) Polarisation  $\sqrt{}$

Remedy – Adding a depolarizer e.g potassium dichromate√

ii) Local action  $\sqrt{}$ 

Remedy – By amalgamation  $\sqrt{}$ 

Accept – use of pure zinc or coating zinc with mercury  $\sqrt{}$