

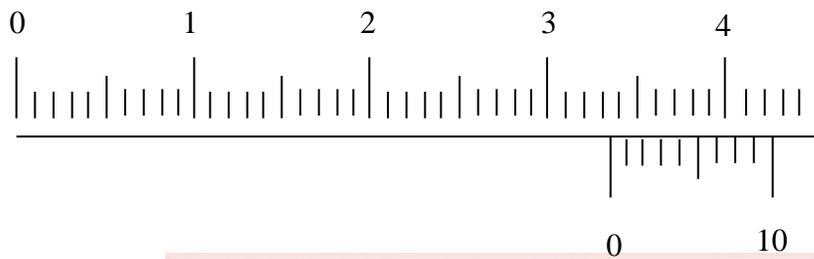
PHYSICS FORM 2

TERM 1 2023

MARKING SCHEME

SECTION A

1.



Check for correct drawing

Main scale 3.3 ✓

Vernier scale 0.06 ✓

2. Volume of water displaced = $100 - 60 = 40\text{cm}^3$

Volume of water displaced = Vol. of stone = 40cm^3 ✓

$$P = \frac{M}{V} \text{ (do not award a mark for the formula)}$$

$$P = \frac{567\text{g}}{40\text{cm}^3} = 14.175\text{g/cm}^3 \text{ (correct substitution) ✓}$$

$P = 14.18\text{g/cm}^3$ (Answer must be given correct to 2d.p)

3. Volume of drop = $5 \times 10^{-8} \text{ M}^3$

i. Area of circular film = 0.1M^2

$$V = A \times H$$

$$h = \frac{V}{A} \checkmark$$

$$\text{Size of molecule} = \frac{5 \times 10^{-8} \text{m}^2}{0.1 \text{m}^2}$$

$$= 5.0 \times 10^{-7} \text{m} \checkmark$$

Accept 0.0000005

Check for correct units.

ii. Atoms are spherical \checkmark

Mass uniformly distributes \checkmark

4. Weight on Earth = 600N

Weight on Planet = 450N

Weight, $W = Mg$

$$M = \frac{W}{g}$$

$$\text{Mass of body} = \frac{600\text{N}}{10\text{N/Kg}} = 60\text{Kg} \checkmark$$

$$g = \frac{W}{m}$$

$$g = \frac{450\text{N} \checkmark}{60\text{Kg}} = \frac{7.5\text{N}}{\text{Kg}} \checkmark$$

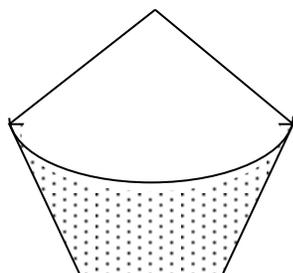
Correct substitution \checkmark

Correct answer with correct units \checkmark

5. The force of cohesion within the mercury is greater than the force of adhesion between mercury and glass ✓. The mercury therefore sinks down ✓the tube to enable mercury molecules to keep together✓.

6. Temperature rise and impurities lower the surface tension of water ✓

7. a)



Check for correct drawing ✓
 Check on the curvature ✓

b) The unbalanced ✓ surface tension✓ pulls the thread tight

8. $h = 760\text{mm}$

$$p = 1.36 \times 10^4 \text{ Kg/m}^3$$

$$p = ?$$

$$p = \rho gh$$

$$p = 1.36 \times 10^4 \times 10 \times \frac{760}{1000}$$

Check on the conversion ✓

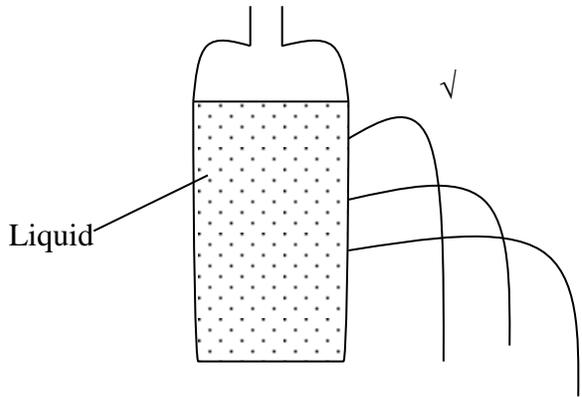
Correct substitution ✓

$$P = 103,360 \text{ N/M}^2$$

Accept $P = 103,360 \text{ pa}$ ✓ check for correct units

9. The external pressure (atmospheric) is lower than the internal pressure ✓: therefore the capillaries break ✓.

10. The bottle with hole experiment – if diagram used; check for labeling✓: Procedure, observation and conclusion✓.



Lowest jet has highest pressure

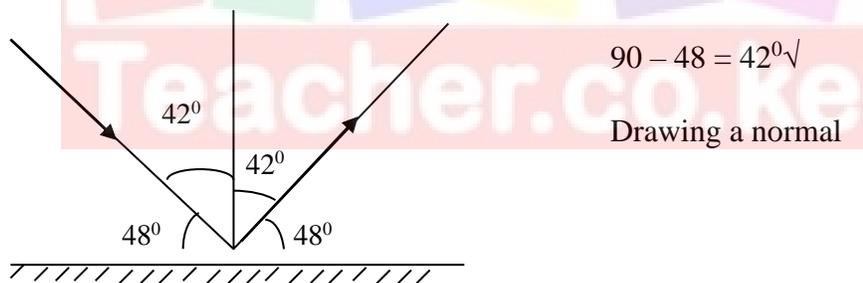
11. Solid – particles very close, hence low kinetic energy ✓

Liquids – particles fairly free, moderate kinetic energy ✓

Gases – particles very free, high kinetic energy ✓

12. The metal blade conducts heat from the hand but the wood cannot ✓

13.



14. $(20 \times 0.3) + (20 \times 0.3)$ ✓ or 20×0.6

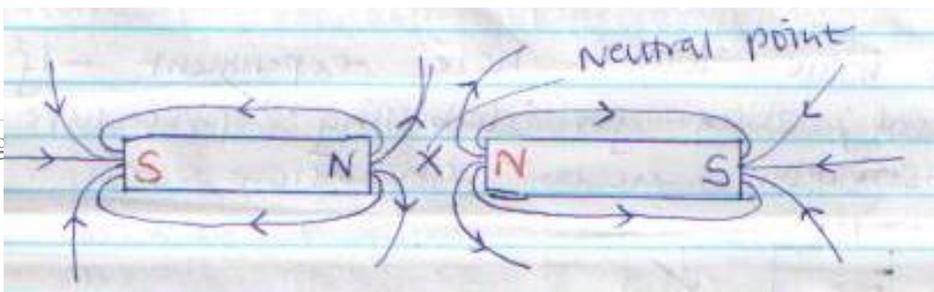
$6 + 6 = 12\text{NM}$ ✓

Check for correct units

15. Unlike poles attract while like poles repel ✓

Reject – unlike charges attract while like charges repel

Reject – unlike terms attract while like terms repel



Check for direction of field ✓

Check for presence of the neutral zone ✓

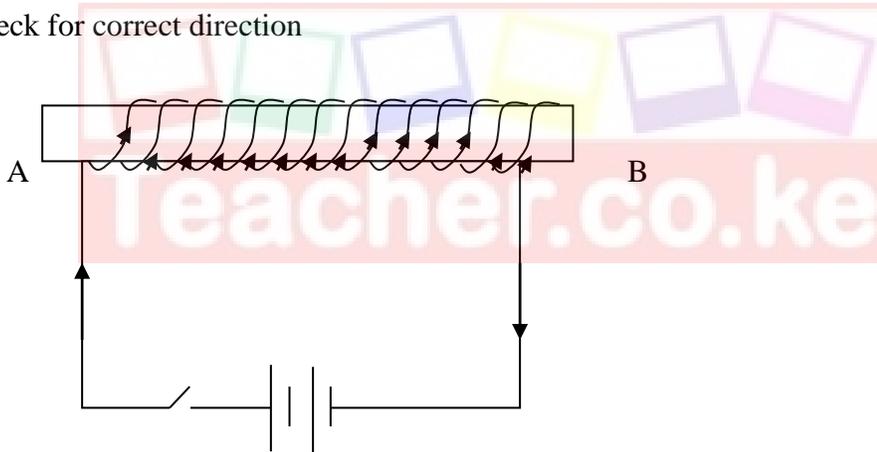
16. This is due to the influence of the Earth's magnetic field ✓

17. Repulsion only occurs between 2 like poles ✓ but attraction may occur between 2 unlike poles or between a magnet and a magnetic materials ✓

SECTION B

18. i. Iron is a soft magnetic material it can easily acquire magnetism and can easily lose magnetism.

ii. Check for correct direction



iii. A – North pole ✓

B – South pole ✓

iv. Right hand grip rule ✓

It states that if a coil carrying current is grasped in the right hand such that the fingers point in the direction of current then the thumb points in the direction of North Pole ✓.

ii. It would cause overheating on the electromagnet√. This adversely affects the magnetism of the electromagnet√.

19. i. Smoke particles – smoke particles are larger than air molecules and light enough to move when bombarded by air molecules √

Lens – focuses the light from the lamp on the smoke particles, causing them to be observable

Microscope – enlarges/magnifies the smoke particles so that they are visible √

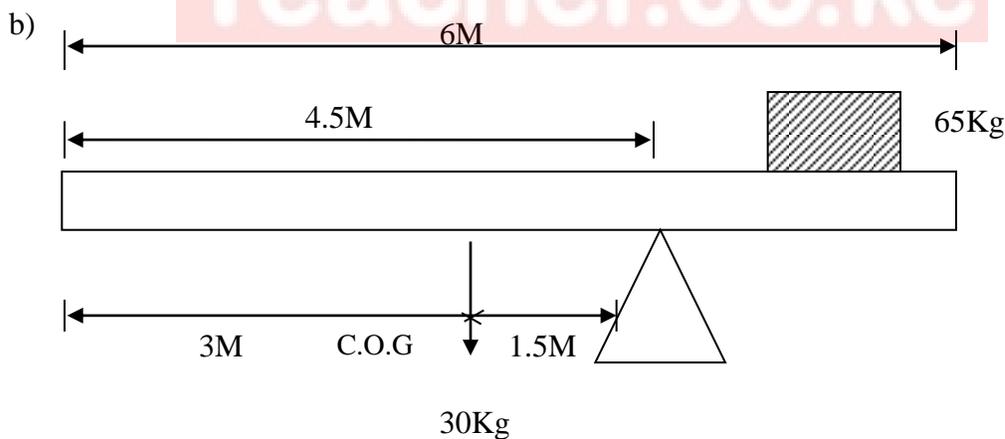
ii. Smoke particles more randomly/zigzag √

Air molecules bombard the smoke particles

Air molecules are in random motion

iii. The speed of motion of smoke particles will be observed to be lighter/faster/speed increases√.

20. a) Principle of moments states that for a system in equilibrium, the sum of clockwise moments must be equal to the sum of the anticlockwise moments.



Clockwise moments = Anticlockwise moments

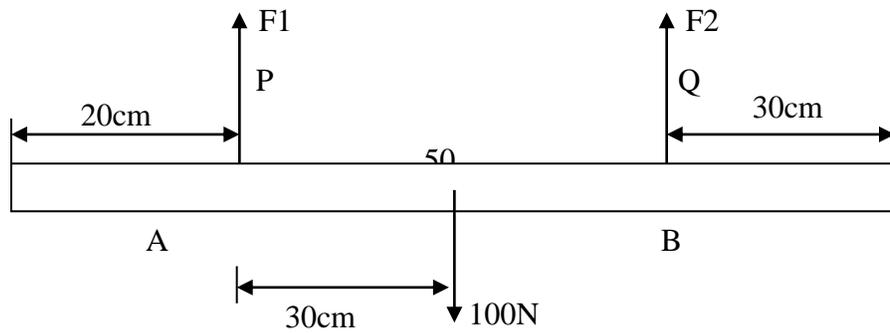
$$300 \times 1.5 = X \times 650 \quad (\text{correct substitution 1mk})$$

$$\frac{450}{650} = \frac{650x}{650}$$

$$X = \frac{450}{650} \sqrt{\quad}$$

$$X = 0.69M \sqrt{\quad}$$

c)



Taking moments about P

$$\text{Distance between P and Q} = 100 - (20 + 30)$$

$$= 100 - 50 \sqrt{\quad}$$

$$= 50 \text{ cm}$$

$$= 0.5 \text{ m}$$

$$F_2 \times 0.5 = 0.3 \times 100 \sqrt{\quad}$$

$$\frac{0.5 F_2}{0.5} = \frac{30}{0.5}$$

$$F_2 = \frac{300}{5} = 60 \text{ N} \sqrt{\quad}$$

Clockwise moments = Anticlockwise moments

$$F_1 + F_2 = 100\text{N}\checkmark$$

$$F_1 + 60\text{N} = 100\text{N}$$

$$F_1 = 100\text{N} - 60\text{N}$$

$$F_1 = 40\text{N}\checkmark$$

21. a) Mass of water = $66.1 - 42.9\checkmark$
 $= 23.2\text{g}\checkmark$

b) Volume = $\frac{\text{Mass}}{\text{Density}} = \frac{23.2\text{g}}{1\text{g/cm}^3}$

$$= 23.2\text{cm}^3\checkmark$$

Working must be shown

c) Volume of density bottle = volume of water

$$\text{Volume of bottle} = 23.2\text{cm}^3\checkmark$$

d) Mass of soil = $67.2 - 42.9$

$$= 24.3\text{g}\checkmark$$

e) Mass of water that filled the space above the soil

$$= 82.0 - 67.2$$

$$= 14.8\text{g}\checkmark$$

f) Volume of soil

$$\text{Volume of water} = \frac{\text{Mass}}{\text{Density}}\checkmark$$

$$= \frac{14.8\text{g}}{1\text{g/cm}^3}$$

$$= 14.8\text{cm}^3\checkmark$$

$$\text{Volume of soil} = 23.2 - 14.8$$

$$= 8.4\text{cm}^3\checkmark$$

g) The density of the soil = $\frac{\text{Mass}}{\text{Volume}}$

$$= \frac{24.3}{8.4}\checkmark$$

$$= 2.893\text{g/cm}^3\checkmark$$

22. a) A – Seal and insulator✓

B – Zinc case✓

C – Mixture of carbon and manganese (IV) oxide✓

D – Carbon rod ✓

b) Zinc case acts as a negative electrode✓

c) i) Polarisation✓

Remedy – Adding a depolarizer e.g potassium dichromate✓

ii) Local action✓

Remedy – By amalgamation✓

Accept – use of pure zinc or coating zinc with mercury ✓