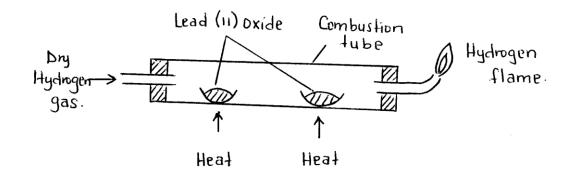
NAME:	•••••
SCHOOL:	•••••
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WATER AND HYDROGEN	
INSTRUCTIONS TO CANDIDATES	
Answer ALL questions in this paper in the spaces provided.	
1. Use the set-up below to answer the questions that follows.	
Lead II oxide	Hydroger burning
Dry \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	
hydrogen ///// gas /////	
a) Why is excess hydrogen burnt at the end of the tube.	(1mk)
b) What type of reaction is occurring in the combustion tube?	(1mk)
c) State one use of hydrogen gas.	(1mk)

2. Study the set up below and use it to answer the questions that follow.



(1)	Why is excess hydrogen burned at the end of the tube?	(Imk)
` '		(1mk)
(iii)) Write a chemical equation for the reaction taking place in the combustion tube.((1mk)